## Quiz Two Key

Version #2

1	$\sim$
- 1	

2. E

4. A

6. D

8.  $(8.44 \times 10^{-3}/3.0L)^2 = 7.9 \times 10^{-6}$ 

The amount of solid doesn't matter as long as there is enough to reach equilibrium.

9. 8.9

10.

- a. No effect. Pure solids and liquids have constant concentration, regardless of total amount.
- b. By a shift in the exothermic direction, the "disturbance" is reduced. More A will form.
- c. More A will form, since this will decrease the amount of gas present and thus reduce the effect of the disturbance.
- d. No effect. Helium does not participate in the reaction, and the added helium will not change the partial pressure of C.